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- (54) Title: CRYSTALLINE AMINE SALT OF CEFDINIR
- (57) Abstract

Cefdinir in the form of a salt with dicyclohexylamine, a process for its production and its use in the purification of impure cefdinir.

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CRYSTALLINE AMINE SALT OF CEFDINIR

The present invention relates to intermediates in the purification of cefdinir, i.e. 7-(Z)-[2-(2-aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid of formula

Cefdinir may be used, e.g. in form of a monohydrate, as a pharmaceutical, e.g. antibiotic; see e.g. Y.Inamoto, Toshiyuki Chiba, Toshiaki Kiamimura und Takao Takaya, J.Antibiotics Vol XLI, No 6, 829, (1988).

Cefdinir may be obtained in impure form according to known production processes. It was now surprisingly found that impure cefdinir may be purified via the formation of a salt, e.g. in crystalline form, thereof.

In one aspect the present invention provides a compound of formula I in the form of a salt, e.g. crystalline, with dicyclohexylamine.

A compound of formula I in the form of a salt with dicyclohexylamine may be produced as follows:

Cefdinir, e.g. in the form of a solvate, such as a hydrate, e.g. in impure form, e.g. as obtained in a production process of cefdinir, such as a mixture of cefdinir and impurities, e.g. such as a mixture of by-products originating from the production process of cefdinir and cefdinir; may be treated in the presence of a solvent, e.g. in dissolved or suspended form, with dicyclohexylamine. A solvent includes any solvent which is inert towards cefdinir or towards cefdinir in the form of a salt with dicyclohexylamine, e.g. a polar organic solvent, such as amides, e.g. dimethylformamide; alcohols, e.g. methanol oder ethanol; ketones, e.g acetone; e.g. in combination with water and water. A solvent system, e.g. mixtures of individual solvents, e.g. as described above may be used. A preferred solvent system may be e.g. acetone/water, including e.g. a ratio of about 100:1 such as 50:1, e.g. 20:1 to 1:5; such as 10:1 to 1:3, e.g. 5:1 to 2:1, e.g. about 1:1. Per equivalent of cefdinir about one equivalent or more, such as 5; e.g. 3, such as 2 equivalents of dicyclohexylamine may be used, e.g. combined

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with the mixture of impure cefdinir in a solvent. A compound of formula I in the form of a salt with dicyclohexylamine may crystallize e.g. from a reaction solution, or, e.g. a suspension of a compound of formula I in a solvent may be converted into a crystall suspension of a compound of formula I in the form of a salt with dicyclohexylamine. An anti-solvent may be added to the reaction mixture, e.g. in order to complete crystallization. An anti-solvent includes solvents wherein a compound of formula I in the form of a salt with dicyclohexylamine is insoluble or soluble only to a small extent if added to the solution or suspension of a compound of formula I in the form of a dicyclohexylamine, e.g. apolar solvents; e.g. ethers, such as diethylether, tetrahydrofurane; or a ketone, e.g. acetone. A compound of formula I in the form of a salt with dicyclohexylamine may be isolated from the reaction mixture, e.g. as conventional, e.g. by filtration, centrifugation.

A compound of formula I in the form of a salt with dicyclohexylamine may be obtained in pure form, e.g. in 98% purity and more, such as 99% to 100% purity; e.g. the amount of impurities present in cefdinir in impure form used for salt formation may be decreased; e.g. impurities of 10% and more in cefdinir in impure form may be decreased to impurities of 1% and less, e.g. 0 to 1% in cefdinir in the form of a salt with dicyclohexylamine.

A compound of formula I in the form of a salt with dicyclohexylamine may be further purified by re-suspension or re-dissolution as described above, e.g. in an (anti) solvent (system), e.g. as described above.

In another aspect the present invention provides a process for the production of a compound of formula I in the form of a salt with dicyclohexylamine, e.g. in crystalline form, comprising treating a compound of formula I, e.g. in form of a solvate, such as a hydrate, in a solvent with dicyclohexylamine and isolating a compound of formula I in the form of a salt with dicyclohexylamine, e.g. in crystalline form.

Cefdinir in free form, e.g. in the form of a solvate, such as a hydrate, e.g. monohydrate and in purified form, e.g. in respect with impure cefdinir used for the formation of a salt of a compound of formula I with dicyclohexylamine, may be obtained from a compound of formula I in the form of a salt with dicyclohexylamine, e.g. as conventional for setting free a compound which is in the form of a salt, e.g. in the form of an amine salt; e.g. by adjusting an appropriate pH, e.g. 1.5 to 4, such as 2 to 3; of a mixture, e.g. a solution, of cefdinir in the form of a salt with dicyclohexylamine with a solvent, e.g. in the presence of water, preferably in water, e.g. by addition of an acidic agent, such as an organic or inorganic acid, preferably

an inorganic acid, e.g. sulphuric acid. Cefdinir, e.g. in the form of hydrate, e.g. monohydrate may crystallize and may be isolated, e.g. as conventional, e.g. by filtration, centrifugation. A compound of formula I may be obtained according to the process of the present invention as such or in the form of a solvate, e.g. a hydrate, e.g. a monohydrate. A compound of formula I obtained according to the process of the present invention as such may be converted into a compound of formula I in the form of a solvate, e.g. a hydrate, such as a monohydrate and vice versa.

In another aspect the present invention provides a process for the production of a compound of formula I, e.g. in form of a solvate, such as a hydrate, e.g. monohydrate comprising converting a compound of formula in the form of a salt with dicyclohexylamine, e.g. in crystalline form, into a compound of formula I, e.g. in the form of a solvate, and isolating a compound of formula I, e.g. in the form of a solvate.

In another aspect the present invention provides a process for the purification of cefdinir in a mixture of a compound of formula I with impurities, comprising forming a salt of a compound of formula I with dicyclohexylamine; and converting a compound of formula I in the form of a salt with dicyclohexylamine, e.g. in crystalline form, into a compound of formula I, e.g. in the form of a solvate, and isolating a compound of formula I, e.g. in the form of a solvate.

A compound of formula I in the form of a salt with dicyclohexylamine is useful in the purification of cefdinir in impure form.

In another aspect the present invention provides the use of a compound of formula I in the form of a salt with dicyclohexylamine, e.g. in crystalline form in the purification of a mixture of a compound of formula I with impurities.

The present invention has several surprising advantages:

A compound of formula I in the form of a salt with dicyclohexylamine may be in acrystalline form; cefdinir in the form of a salt may be obtained in surprising high purity, e.g. 98% purity and more, e.g. 98% to 100%; production of the salt is simple; cefdinir obtained from the salt may be surprisingly pure, e.g. 98% and more, e.g. 99% to 100%.

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It is surprising that cefdinir under the basic conditions of the salt formation according to the present invention is stable, because from e.g. Yoshihiko Okamoto et al., J. of Pharmaceutical Sciences, Vol 85, No 9, 976, (1996) it is known that cefdinir may be instable in a basic environment; it was e.g. found that cefdinir in the presence of other amines, e.g. tert.-octylamine under similar conditions may be, e.g. heavily, degraded.

In another aspect the present invention provides crystalline 7-(Z)-[2-(2-aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid.dicyclohexylammonium salt of formula

In the following examples, which illustrate the invention more fully but do in no way limit its scope, all temperatures are given in degrees Celsius. Purity of a compound obtained is determined by HPLC.

Example 1

7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid in form of a salt with dicyclohexylamine

10 g of crude cefdinir, e.g. as obtained in a cefdinir production process, in 50 ml of water and 50 ml of acetone are treated under stirring with 5 ml of dicyclohexylamine. A solution is obtained and 7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyimino-acetamido]-3-vinyl-3-cephem-4-carboxylic acid in form of a salt with dicyclohexylamine crystallizes. 250 ml of acetone are added to the crystall suspension which is stirred for ca. 30 minutes at room temperature. The crystalls are filtrated off, washed with acetone and dried. Crystalline 7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid in form of a salt with dicyclohexylamine in a purity of 98.6 % is obtained. Mp: 175° (decomposition).

¹H-NMR (DMSO-d6): 9.41 (d, 1H, J = 8.1 Hz, NH); 7.12 (s, 2H, NH₂); 6.99 (dd, 1H, J = 11.4 and 17.7 Hz, CH = CH₂); 6.64 (s, 1H, thiazol); 5.60 (dd, 1H, J = 4.8 and 8.1 Hz, H₇); 5.15 (d, 1H J = 17.7 Hz, CH = CH₂); 5.04 (d, 1H, J = 4.8 Hz, H₆); 4.94 (d, 1H, J = 11.4 Hz, CH = CH₂); 3.52, 3.39 (AB d, 1H, J = 17 Hz, H₂); 3.21 (m, 2H); 2.05 (m, 4H); 1.8 (m, 4H); 1.6 (m, 2H); 1.2 - 1.4 (m, 10H).

Example 2

7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid in the form of a monohydrate

10 g of 7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid in form of a salt with dicyclohexylamine, obtained according to Example 1 are dissolved in 175 ml of water at a temperature of ca. 35-40° and treated with active charcoal. Active charcoal is filtrated off and the pH of the solution obtained is adjusted to pH 2.5 by addition of 5ml of sulphuric acid at ca. 35°. 7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyimino-acetamido]-3-vinyl-3-cephem-4-carboxylic acid in the form of a monohydrate precipitates, is filtrated off, washed with water and dried. 7-(Z)-[2-(2-Aminothiazol-4-yl)-2-hydroxyimino-acetamido]-3-vinyl-3-cephem-4-carboxylic acid in the form of a monohydrate in a purity of 99% is obtained.

Example 3

Production of crude cefdinir

40 g of 7-amino-3-vinyl-3-cephem-4-carboxylic acid in 400 ml of dichloromethane are treated with 55.7 ml of N,O-bistrimethyl-silylacetamid. The mixture obtained is stirred for ca. 2 hours at room temperature, cooled to 0° and treated with 52.2 g of 2-(Z)-(2-aminothiazol-4-

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yl)-2-acetoxyiminoacetic acid chloride in the form of a hydrochloride in small portions. The mixture obtained is stirred for ca.90 minutes at 0° and added under stirring to a mixture of 44.55 g of NaHCO3, 600 ml of water und 100 ml of dichloromethane of a temperature of 5°. The pH of the mixture obtained is adjusted to a pH of ca. 7.2 - 7.3 with a saturated aqueous solution of NaHCO3. The phases formed are separated. To the aqueous phase 300 ml of water are added and 28.7 g of NH4Cl. The pH of the mixture obtained is adjusted to pH 8 by addition of an aqueous 10% K2CO3 solution and the mixture is stirred for ca. 80 minutes. A solution is obtained. The pH of the solution obtained is adjusted to pH 3 by addition of 5m sulphuric acid. 7-(Z)-[2-(2-amino-thiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid precipitates, is filtrated off, washed with water and dried. 7-(Z)-[2-(2-amino-thiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid in a purity of 94.3 % is obtained.

Patent claims

1. A compound of formula

in the form of a salt with dicyclohexylamine.

- 2. A process for the production of a compound of formula I as defined in claim 1 in the form of a salt with dicyclohexylamine comprising treating a compound of formula I in a solvent with dicyclohexylamine and isolating a compound of formula I in the form of a salt with dicyclohexylamine.
- 3. A process for the production of a compound of formula I as defined in claim 1 comprising converting a compound of formula I in the form of a salt with dicyclohexylamine into a compound of formula I and isolating a compound of formula I.
- 4. A process for the purification of a compound of formula I as defined in claims 1 in a mixture of a compound of formula I with impurities, comprising forming a salt of a compound of formula I with dicyclohexylamine; and converting a compound of formula I in the form of a salt with dicyclohexylamine into a compound of formula I and isolating a compound of formula I.
- 5. A process according to any one of claims 3 to 4 comprising converting a compound of formula I in the form of a salt with dicyclohexylamine into a compound of formula I in the form of a solvate.
- 6. A process according to any one of claims 3 to 5 comprising isolating a compound of formula I in the form of a solvate.
- 7. Use of a compound of claim 1 in the purification of a mixture of a compound of formula I with impurities.

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- 8. A compound of formula I as defined in claim I in the form of a salt with dicyclohexylamine according to any one of claims 1 to 7, which is in crystalline form.
- 9. Crystalline 7-(Z)-[2-(2-aminothiazol-4-yl)-2-hydroxyiminoacetamido]-3-vinyl-3-cephem-4-carboxylic acid.dicyclohexylammonium salt of formula

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Electronic o	data base consulted during the international search (name of da	ata base and, where practical, se	earch terms used)
C DOCUM	TENTS CONCIDENTS TO SEE THE		
Category °	ENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the		
	where appropriate, or the	e relevani passages	Relevant to claim No.
Χ	US 4 559 334 A (T. TAKAYA ET A	AL.) 17	1,8,9
	December 1985	·	1,0,5
	see column 2, line 5; claims		
Υ	WO 97 07121 A (BIOCHEMIE GMBH)	27 February	1-9
	1997 see page 11 - page 12; claims		
Υ	GB 1 038 529 A (FUJISAWA PHARM	MACEUTICAL	1-9
	CO., LTD.) 10 August 1966 see page 10, example 32; claim	1 33	
А	EP 0 304 019 A (FUJISAWA PHARM CO., LTD.) 22 February 1989	1ACEUTICAL	1-9
	see claims		
		,	
		-/	
X Furti	her documents are listed in the continuation of box C.	χ Patent family me	mbers are listed in annex.
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Category '	Citation of document, with indication where appropriate of the relevant		
cgviy	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.	
, А	WO 97 24358 A (HANMI PHARMACEUTICAL CO., LTD.) 10 July 1997 see claims	1-9	
			4
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INTERNATIONAL SEARCH REPORT

normation on patent family members

Ir. ational Application No PCT/EP 98/01953

	ent document		Publication		Patent family	Publication
	n search repor	t ————	date		member(s)	date
US 4	4559334	Α	17-12-1985	AT	385994 B	10-06-1988
				AU	576735 B	08-09-1988
				AU	1927783 A	05-04-1984
				CA	1206956 A	01-07-1986
				CH	657857 A	30-09-1986
				DK	427083 A,B,	31-03-1984
				EP	0105459 A´´	18-04-1984
				FI	833370 A,B,	31-03-1984
				FR	2533926 A	06-04-1984
				GB	2127812 A,B	18-04-1984
				JP	1926846 C	25-04-1995
				JP	6057713 B	03-08-1994
				JP	62294687 A	22-12-1987
				PT	77426 B	27-02-1986
				SU	1309911 A	07-05-1987
WO 9	9707121	Α	27-02-1997	AT	403049 B	27-10-1997
				AT	136995 A	15-03-1997
				AU	6821396 A	12-03-1997
				EP	0844999 A	03-06-1998
GB 1	1038529	Α		DE	1545794 A	11-12-1969
				FR	5564 M	02-01-1968
				SE	329165 B	05-10-1970
EP 3	304019	Α	22-02-1989	AT	123221 T	15-06-1995
				AU	617347 B	28-11-1991
				CA	1297096 A	10-03-1992
				DE	3853901 D	06-07-1995
				DE	3853901 T	12-10-1995
				ES	2072856 T	01-08-1995
				HK	18496 A	09-02-1996
				IE	67348 B	20-03-1996
				JP	1250384 A	05-10-1989
				JP	1943842 C	23-06-1995
				JP	6074276 B	21-09-1994
				KR	9708126 B	21-05-1997
				MX	9203468 A	01-09-1992
				US	4935507 A	19-06-1990

Form PCT/ISA/210 (patent family annex) (July 1992)

Information on patent family members

ational Application No

			action on patent raising men	idera	PCT/EP	98/01953			
Pater cited in	nt document search report		Publication date	Patent family member(s)	/	Publication date			
WO 9	724358	Α	10-07-1997	NONE	<u> </u>				
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Form PCT/ISA/210 (patent family annex) (July 1992)